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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/677,478	10/02/2000	Guy T. Blalock	M122-1544	4522

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EXAMINER

VINH, LAN

ART UNIT	PAPER NUMBER
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1765

DATE MAILED: 03/29/2002

6

Please find below and/or attached an Office communication concerning this application or proceeding.

T.D-6

Office Action Summary

Application No.

09/677,478

Applicant(s)

BLALOCK ET AL.

Examiner

LAN VINH

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-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 02 October 2000.
- 2a) ☐ This action is FINAL. 2b) ☒ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-7, 10-13, 16-28, 30-42, 44-50 and 53-57 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-7, 10-13, 16-28, 30-42, 44-50 and 53-57 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
- 11) ☐ The proposed drawing correction filed on _____ is: a) ☐ approved b) ☐ disapproved by the Examiner.
If approved, corrected drawings are required in reply to this Office action.
- 12) ☐ The oath or declaration is objected to by the Examiner.

Priority under 35 U.S.C. §§ 119 and 120

- 13) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. _____.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).
* See the attached detailed Office action for a list of the certified copies not received.
- 14) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).
a) ☐ The translation of the foreign language provisional application has been received.
- 15) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

Attachment(s)

- 1) ☒ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☒ Information Disclosure Statement(s) (PTO-1449) Paper No(s) 3, 5.
- 4) ☐ Interview Summary (PTO-413) Paper No(s). _____
- 5) ☐ Notice of Informal Patent Application (PTO-152)
- 6) ☐ Other:

DETAILED ACTION

Specification

1. The applicants are requested to provide the US patent number under the related patent data section in the specification.

Claim Rejections - 35 USC § 103

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Claims 1-5, 7 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kilgore et al (US 6,200,412) in view of Keller et al (US 5,644,153)

Kilgore discloses a plasma etching method comprising the steps of:

flowing fluorocarbons gas into the chamber to form residual deposits of fluorine/halogen over the surface of the reaction chamber (col 6, lines 11-12) reads on forming residual deposit of halogen over at least some internal surfaces of a plasma etch chamber

subsequently, plasma etching using hydrogen plasma to remove/etch the fluorine residual deposits from the surfaces of the reaction chamber, the hydrogen component reacts with the fluorine/halogen to form gaseous HF (hydrogen halide) (col 6, lines 18-22) reads on after forming the residual deposits, plasma etching using a gas effective to etch the residual fluorine deposits from chamber internal surfaces, the gas having a

hydrogen component effective to form a gaseous hydrogen halide from halogen/fluorine liberated from the polymer

Unlike the instant claimed invention as per claim 1, Kilgore does not specifically disclose forming a polymer comprising carbon and a halogen over the internal surface of a plasma chamber although Kilgore discloses forming residual deposits of fluorine/halogen over the surface of the reaction chamber.

However, Keller, in a plasma etching method using fluorocarbon gases, discloses that residual reaction deposits inside the plasma chamber are polymeric deposits comprises of carbon and fluorine/halogen (col 6, lines 31-32; col 7, lines 48-49)

Since both Kilgore and Keller are concerned with plasma etching method using fluorocarbon gases to form residual deposit in the chamber, one skilled in the art would have found it obvious that Kilgore etching step using fluorocarbon gas would have resulted in forming a polymeric residual deposit comprising carbon and a halogen in view of Keller teaching especially since Keller states that the polymer will typically be comprises of carbon and fluorine (halogen) (col 6, lines 30-31)

The limitation of fluorine as the claimed halogen as recited in claims 2 and 3 has been discussed above.

Regarding claim 4 and 5, Kilgore discloses that oxygen can be used with the hydrogen plasma (col 6, lines 42-43)

The limitation of claim 7 has been discussed above in paragraph 3.

4. Claim 6 is rejected under 35 U.S.C. 103(a) as being unpatentable over Kilgore et al (US 6,200,412) in view of Keller et al (US 5,644,153) and further in view of William et al (US 5,647,953)

Kilgore as modified by Keller has been described above in paragraph 3. Kilgore and Keller differ from the instant claimed invention as per claim 6 by using hydrogen component instead of ammonia.

However, William discloses that fluorine residues/polymer can be removed by a reducing gas such as hydrogen, ammonia (col 2, lines 61-63)

Hence, one skilled in the art would have found it obvious to substitute Kilgore's hydrogen plasma to remove /etch residue with ammonia in view of William teaching because both hydrogen and ammonia are known reducing gas having the same function of removing fluorine residue; thus, the substitution of one for the other would have been anticipated to produce an expected result.

5. Claims 10-13, 16-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Cui et al (US 5,965,463) in view of Yanagida (US 5,445,712)

Cui discloses a plasma etching method comprising the steps of:

forming a polymer comprises of carbon and fluorine/halogen on all surfaces/internal surface of a plasma chamber (col 2, lines 4-11)

performing a post etch plasma etching using a gas to remove/etch the polymer from the chamber surface (col 10, lines 24-25) reads on after forming the polymer, plasma etching using a gas effective to etch polymer from the chamber internal surfaces. Cui

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also discloses using a gas C_4F_8 / the gas comprises a carbon compound to remove polymer from chamber surface.

Unlike the instant claimed invention as per claim 10, Cui does not specifically discloses the carbon compound effective to getter/to remove halogen from the etched polymer.

However, Yanagida discloses a plasma etching method comprises the step of using a carbon compound gas to form gaseous HF / to form a gaseous hydrogen halide from the etched fluorine/halogen (col 2, lines 25-29)

Since both Cui and Yanagida are concerned with method of plasma etching using carbon compound gas, one skilled in the art would have found it obvious to employ Cui carbon compound gas to getter/ to remove the halogen from the etched polymer as per Yanagida because Yanagida states that C atoms combined with halogen radicals dissociated from the halogen compound and is quickly removed in the form of halide (col 4, lines 6-10)

The limitation of claim 11 has been discussed above.

Regarding claim 12, since Cui discloses using CO gas during the etching process, one skilled in the art would have obvious that Cui's CO gas would have formed a gaseous COF in view of Yanagida teaching so the gaseous COF can be removed from the chamber.

Regarding claim 13, Cui discloses using CHF_3 / hydrocarbon gas (col 11, lines 13-14)

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Regarding claims 16-18, Cui discloses flowing CO gas into the chamber (col 9, lines 24-25)

The limitation of claim 19 has been discussed above in paragraph 5.

Regarding claim 20, Cui discloses using oxygen plasma (col 10, lines 30-31)

6. Claims 21-26, 28, 30 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kilgore et al (US 6,200,412) in view of Keller et al (US 5,644,153) and further in view of Saito et al (5,681,424)

Kilgore discloses a plasma etching method comprising the steps of:

resting/positioning a semiconductor wafer on a platen (claimed wafer receiver) within a chamber (col 56-58)

flowing fluorocarbons gas into the chamber to remove/etch material on the wafer and to form residual deposits of fluorine/halogen over the surface of the reaction chamber (col 6, lines 11-12) reads on first etching material on the wafer with a gas comprising carbon and halogen to form residual deposit of halogen over at least some internal surfaces of a plasma etch chamber

subsequently, with the wafer on the platen/wafer receiver, plasma etching using hydrogen plasma to remove/etch the fluorine residual deposits from the surfaces of the reaction chamber, the hydrogen component reacts/getter with the fluorine/halogen to form gaseous HF (hydrogen halide) (col 6, lines 18-22) reads on after the first plasma, plasma etching using a gas effective to etch the residual fluorine deposits from

chamber internal surfaces, the gas having a hydrogen component effective to form a gaseous hydrogen halide from halogen/fluorine liberated from the polymer

Unlike the instant claimed invention as per claim 21, Kilgore does not specifically disclose forming a polymer comprising carbon and a halogen over the internal surface of a plasma chamber although Kilgore discloses forming residual deposits of fluorine/halogen over the surface of the reaction chamber.

However, Keller, in a plasma etching method using fluorocarbon gases, discloses that residual reaction deposits inside the plasma chamber are polymeric deposits comprises of carbon and fluorine/halogen (col 6, lines 31-32; col 7, lines 48-49)

Since both Kilgore and Keller are concerned with plasma etching method using fluorocarbon gases to form residual deposit in the chamber, one skilled in the art would have found it obvious that Kilgore etching step using fluorocarbon gas would have resulted in forming a polymeric residual deposit comprising carbon and a halogen in view of Keller teaching especially since Keller states that the polymer will typically be comprises of carbon and fluorine (halogen) (col 6, lines 30-31)

Kilgore and Keller do not disclose etching the polymer from chamber internal surface to restrict further etching of the material during the second plasma etching.

Saito discloses a plasma etching method comprises the step of using a cleaning gas to remove the polymer on the chamber surface and to suppress/restrict further etching of the material (oxide) on the wafer (col 1, lines 63-65 and col 4, lines 53-54)

Hence, one skilled in the art would have found it obvious to employ Kilgore and Keller cleaning gas to reduce further etching of the material in view of Saito teaching in order to reduce etching time by eliminating the overetching time (col 5, lines 6-8)

Regarding claim 22, one skilled in the art would have found it obvious to provide power at floating potential or ground potential to Kilgore platten/wafer receiver (biasing) in view of Saito teaching (col 3, lines 3-5) to enhance cooling efficiency during etching (col 3, lines 8-9)

The limitation of claims 23, 25, 28 has been discussed above in paragraph 6.

Regarding claim 24, Kilgore discloses maintaining the reaction chamber components at the temperature of 25-150° C (col 6, lines 24-25) reads on providing a receiver at temperature from -10 to about 40° C and keeping the chamber pressure at 0.1 to 5 Torr (col 6, lines 44-45) reads on the claimed range of 30 mTorr-5 Torr

Regarding claim 30, Kilgore discloses performing the etching and the cleaning step in a same chamber (in situ), creating high-density plasma at low pressure (0.1 -100 mTorr) (col 5, lines 23-24; col 6, lines 1-3) reads on maintaining the chamber at sub atmospheric pressure.

7. Claim 27 is rejected under 35 U.S.C. 103(a) as being unpatentable over Kilgore et al (US 6,200,412) in view of Keller et al (US 5,644,153) and Saito et al (US 5,681,424) and further in view of William et al (US 5,647,953)

Kilgore as modified by Keller and Saito has been described above in paragraph 6. Kilgore, Keller and Saito differ from the instant claimed invention as per claim 6 by using hydrogen component instead of ammonia.

However, William discloses that fluorine residues/polymer can be removed by a reducing gas such as hydrogen, ammonia (col 2, lines 61-63)

Hence, one skilled in the art would have found it obvious to substitute Kilgore, Keller and Saito hydrogen plasma to remove /etch residue with ammonia in view of William teaching because both hydrogen and ammonia are known reducing gas having the same function of removing fluorine residue; thus, the substitution of one for the other would have been anticipated to produce an expected result.

8. Claims 31-35 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kilgore et al (US 6,200,412) in view of Keller et al (US 5,644,153) and Saito et al (5,681,424) and further in view of Yanagida (US 5,445,712)

Kilgore as modified by Keller and Saito has been described above in paragraph 6.

Kilgore, Keller and Saito differ from the instant claimed inventions as per claims 31-33 by gettering/removing the etched polymer using hydrogen instead of a carbon compound

However, Yanagida discloses a plasma etching method comprises the step of using a carbon compound (C-O bond) gas to form gaseous HF / to form a gaseous hydrogen halide from the etched fluorine/halogen (to getter halogen from the etched fluorine) (col 2, lines 25-29)

Hence, one skilled in the art would have found it obvious to modify Kilgore, Keller and Saito by using a carbon compound (C-O bond) to getter halogen from the etched fluorine as per Yanagida because hydrogen plasma and carbon compound plasma have the same function of removing/gettering halogen from the etched polymer; thus the substitution of one for the other would have been anticipated to produce an expected result.

9. Claims 36-42, 44-50, 53 are rejected under 35 U.S.C. 103(a) as being unpatentable over Cui et al (US 5,965,463) in view of Kilgore et al (US 6,200,412) and further in view of Saito et al. (US 5,681,424)

Cui discloses a plasma etching method comprising the steps of:

positioning a semiconductor wafer on a pedestal/electrostatic chuck within an inductively coupled plasma etch chamber (col 6, lines 32-36), the wafer having a photoresist layer 18 formed on an insulating layer 16 (oxide), the photoresist layer having opening patterns (fig. 5)

using the patterned photoresist as a mask to etch opening within the insulating layer 16 with fluorocarbons gases (carbon and fluorine) to form a polymer comprises of carbon and fluorine/halogen on all surfaces/internal surface of a plasma chamber (col 2, lines 4-11;col 5, lines 59-61)

subsequently, with the wafer on the pedestal/electrostatic chuck at biasing power (floating potential) (col 10, lines 30-33) using oxygen gas/oxygen component in a

second plasma to etch the photoresist and to remove/etch the polymer from the chamber surface (col 5, lines 60-61; col 10, lines 24-25)

Cui differs from the instant claimed invention as per claims 36, 47 by performing the second plasma using oxygen component instead of a gas comprising an oxygen and a hydrogen component to form HF from fluorine (halogen) liberated from the polymer/getter halogen liberated from the polymer.

However, Kilgore discloses a plasma etching method comprising the step of: plasma etching using hydrogen plasma with oxygen to remove/etch the fluorine residual deposits from the surfaces of the reaction chamber, the hydrogen component reacts/getter with the fluorine/halogen to form gaseous HF (hydrogen halide) (col 6, lines 42-44)

Since both Cui and Kilgore are concerned with method of cleaning residue/polymer from the chamber surface, one skilled in the art would have found it obvious to modify Cui method by using hydrogen plasma with oxygen component to remove/etch the fluorine residual and to form HF as per Kilgore because Kilgore teaches that HF can be pumped away through the exhaust system of the chamber (col 6, lines 20-22)

Cui and Kilgore do not disclose etching the polymer from chamber internal surface to restrict further etching of the material/ widening of the opening during the second plasma etching.

Saito discloses a plasma etching method comprises the step of using a cleaning gas to remove the polymer on the chamber surface and to suppress/restrict further etching of the material (oxide) on the wafer (col 1, lines 63-65 and col 4, lines 53-54)

Hence, one skilled in the art would have found it obvious to employ Cui and Kilgore cleaning gas to reduce further etching of the material in view of Saito teaching in order to reduce etching time by eliminating the overetching time (col 5, lines 6-8)

The limitations as recited in claims 37, 38, 40, 42, 48, 50 have been discussed above in paragraph 9.

Regarding claim 39, Cui, Kilgore and Saito do not disclose the specific volumetric ratio/etchant concentration ratio. However, since it is known in the art of plasma etching that etching parameter such as etchant concentration affects the etching rate (see prior art of record for evidence of this basis), it would be obvious to adjust the etchant concentration by optimizing the same by conducting routine experimentation for the purpose of obtaining desirable etch rate.

Regarding claims 41 and 49, the limitations of using ammonia as a hydrogen component has been discussed above in paragraph 4.

Regarding claims 44, 53, Cui discloses performing the etching and cleaning step in a same chamber (in situ) while maintaining the chamber pressure at 35-45 mTorr (subatmospheric pressure) (col 6, lines 33-35; col 10, lines 25-33)

Regarding claims 45-46, Cui discloses flowing CO gas (carbon compound) into the chamber (col 9, lines 24-25)

10. Claims 54-57 are rejected under 35 U.S.C. 103(a) as being unpatentable over Cui et al (US 5,965,463) in view of Yanagida (US 5,445,712) and further in view of Saito et al. (US 5,681,424)

Cui method has been described above in paragraph 9. Cui differs from the instant claimed invention as per claim 54 by performing the second plasma using oxygen component instead of a gas comprising an oxygen and a carbon component to etch photoresist from the substrate and polymer from the chamber surface and gettering fluorine liberated from the polymer.

However, Yanagida discloses a plasma etching method comprises the step of using a carbon compound gas to form a gaseous hydrogen halide from the etched fluorine/halogen/ to getter fluorine liberated from the etched polymer (col 2, lines 25-29)

Since both Cui and Yanagida are concerned with method of plasma etching and Cui also mentions using a gas comprises a carbon compound to remove polymer from chamber surface (col 10, lines 28-29), one skilled in the art would have found it obvious to employ Cui carbon compound gas to getter/ to remove the halogen from the etched polymer as per Yanagida because Yanagida states that C atoms combined with halogen radicals dissociated from the halogen compound and is quickly removed in the form of halide (col 4, lines 6-10)

Cui and Yanagida do not disclose etching the polymer from chamber internal surface to restrict further etching of the material/ widening of the opening during the second plasma etching.

Saito discloses a plasma etching method comprises the step of using a cleaning gas to remove the polymer on the chamber surface and to suppress/restrict further etching of the material (oxide) on the wafer (col 1, lines 63-65 and col 4, lines 53-54)

Hence, one skilled in the art would have found it obvious to employ Cui and Yanagida cleaning gas to reduce further etching of the material in view of Saito teaching in order to reduce etching time by eliminating the overetching time (col 5, lines 6-8)

The limitation of claim 55 has been described above in paragraph 10.

Regarding claims 56, 57, since Cui discloses using CO gas during the etching process, one skilled in the art would have obvious that Cui's CO gas would have formed a gaseous COF in view of Yanagida teaching so the gaseous COF can be removed from the chamber.

11. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure.

Clampitt (US 5,798,303) discloses that etching parameter such as etchant concentration may be varied to achieve desired etch rate (col 6, lines 29-31)

Response to Arguments

12. Applicant's arguments with respect to claims 1-7, 10-13, 16-28, 30-42, 44-50, 53-57 have been considered but are moot in view of the new ground(s) of rejection.

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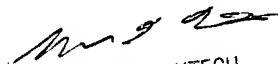
Conclusion

13. Any inquiry concerning this communication or earlier communications from the examiner should be directed to LAN VINH whose telephone number is 703 305-6302. The examiner can normally be reached on Monday-Friday 8:30 -6:00 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, BENJAMIN L UTECH can be reached on 703 308-3836. The fax phone numbers for the organization where this application or proceeding is assigned are 703 872-9310 for regular communications and 703 872-9311 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703 308-0661.

LV
March 24, 2002


BENJAMIN L. UTECH
SUPERVISORY/TECHNICAL EXAMINER
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